

2-[2-(1,3-Dioxoisindolin-2-yl)phenyl]-isoindoline-1,3-dione

Xiao-Yan Wu, Dong-Sheng Zhu,* Yan-Hua Wang and Chao-Guang Zhu

Department of Chemistry, Northeast Normal University, Changchun 130024, People's Republic of China

Correspondence e-mail: zhuds206@nenu.edu.cn

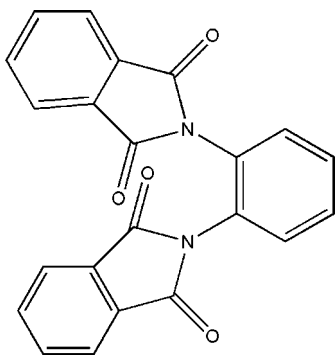
Received 21 July 2007; accepted 28 July 2007

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.041; wR factor = 0.094; data-to-parameter ratio = 13.7.

In the title compound, $\text{C}_{22}\text{H}_{12}\text{N}_2\text{O}_4$, the two isoindolinyl groups are twisted with respect to the benzene ring, with dihedral angles of 58.15 (7) and 61.32 (7)°; the dihedral angle between the two isoindolinyl groups is 39.45 (6)°. The molecules are linked together by weak $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonding.

Related literature

For general background, see: Pendrak *et al.* (1994); De Clerck (1995); Stowers (1996); Heaney & Shuhaibar (1995). For a related structure, see: Liang & Li (2007).



Experimental

Crystal data

$\text{C}_{22}\text{H}_{12}\text{N}_2\text{O}_4$
 $M_r = 368.34$

Monoclinic, $P2_1/n$
 $a = 10.7765$ (10) Å

$b = 16.9259$ (17) Å
 $c = 10.8518$ (10) Å
 $\beta = 116.780$ (1)°
 $V = 1767.1$ (3) Å³
 $Z = 4$

Mo $K\alpha$ radiation
 $\mu = 0.10$ mm⁻¹
 $T = 293$ (2) K
 $0.35 \times 0.33 \times 0.28$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: none
9771 measured reflections

3476 independent reflections
2553 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.050$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.094$
 $S = 0.95$
3476 reflections

253 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.19$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C13}-\text{H13A}\cdots\text{O4}^{\text{iv}}$	0.93	2.39	3.321 (2)	179
$\text{C24}-\text{H24A}\cdots\text{O3}^{\text{iii}}$	0.93	2.51	3.405 (2)	161
$\text{C26}-\text{H26A}\cdots\text{O1}^{\text{iii}}$	0.93	2.53	3.434 (2)	164
$\text{C34}-\text{H34A}\cdots\text{O2}^{\text{iv}}$	0.93	2.48	3.410 (2)	177
$\text{C36}-\text{H36A}\cdots\text{O2}^{\text{v}}$	0.93	2.55	3.255 (2)	133

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$; (ii) $x - 1, y, z - 1$; (iii) $-x + 1, -y, -z$; (iv) $x + \frac{1}{2}, -y + \frac{1}{2}, z - \frac{1}{2}$; (v) $x + 1, y, z$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1999); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL-Plus* (Sheldrick, 1990); software used to prepare material for publication: *SHELXTL-Plus*.

This work was supported by the Postdoctoral Science Foundation, China (grant No. 2005038561)

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2298).

References

- Bruker (1997). *SMART*. Version 5.622. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (1999). *SAINTE*. Version 6.02. Bruker AXS Inc., Madison, Wisconsin, USA.
 De Clerck, E. (1995). *J. Med. Chem.* **38**, 2491–2517.
 Heaney, H. & Shuhaibar, K. F. (1995). *Synlett*, pp. 47–49.
 Liang, Z.-P. & Li, J. (2007). *Acta Cryst.* **E63**, o405–o406.
 Pendrak, I., Barney, S., Wittrock, R., Lambert, D. M. & Kingsbury, W. E. (1994). *J. Org. Chem.* **59**, 2623–2625.
 Sheldrick, G. M. (1990). *SHELXTL-Plus*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
 Stowers, J. R. (1996). *Tetrahedron*, **52**, 3339–3354.

supplementary materials

Acta Cryst. (2007). E63, o3694 [doi:10.1107/S1600536807037038]

2-[2-(1,3-Dioxoisindolin-2-yl)phenyl]isoindoline-1,3-dione

X.-Y. Wu, D.-S. Zhu, Y.-H. Wang and C.-G. Zhu

Comment

Isoindolinones and their derivatives have been investigated widely due to their profound physiological and chemotherapeutic properties. Many compounds containing the isoindolinone skeleton have shown antiviral, antileukemic, antiinflammatory, antipsychotic and antiulcer properties (Pendrak *et al.*, 1994; De Clerck, 1995). Isoindolinones are useful for the synthesis of various drugs and naturally occurring compounds (Stowers, 1996; Heaney & Shuhaibar, 1995). In this paper, we report the crystal structure of the title compound.

As shown in Fig. 1, the structure determination indicates that the molecule is non-centrosymmetric. Due to steric encumbering, two isoindolinyl moieties are twisted to the benzene ring with dihedral angles of 58.15 (7) and 61.32 (7)°, respectively, similar to 62.31 (6)° found in 2-(2-pyridyl)isoindoline-1,3-dione (Liang & Li, 2007). The dihedral angle between the two isoindolinyl moieties is 39.45 (6)°. The molecules are linked together by weak C—H···O hydrogen bonding (Table 1).

Experimental

O-Phenylenediamine (0.11 g, 1 mmol) was dissolved in 5 ml glacial acetic acid, and a solution of phthalic anhydride (0.148 g, 1 mmol) in 10 ml of glacial acetic acid was added. After 30 min, the mixture was diluted with cool water, and then the grey precipitate was separated and then refluxed in dry benzene (20 ml) for 4 h using Dean and Stark apparatus. After solvent was removed under reduced pressure the solid product was obtained. The colorless single crystals of the title compound were obtained by recrystallization from an ethanol solution at room temperature after 2 d.

Refinement

H atoms were positioned geometrically and refined as riding atoms, with C—H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Figures

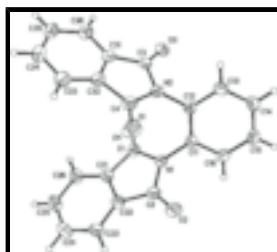


Fig. 1. A view of the molecule of (I). Displacement ellipsoids are drawn at the 50% probability level.

2-[2-(1,3-Dioxoisindolin-2-yl)phenyl]isoindoline-1,3-dione

Crystal data

$C_{22}H_{12}N_2O_4$	$F_{000} = 760$
$M_r = 368.34$	$D_x = 1.385 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
Hall symbol: -P 2yn	$\lambda = 0.71073 \text{ \AA}$
$a = 10.7765 (10) \text{ \AA}$	Cell parameters from 9771 reflections
$b = 16.9259 (17) \text{ \AA}$	$\theta = 2.2\text{--}26.0^\circ$
$c = 10.8518 (10) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$\beta = 116.780 (1)^\circ$	$T = 293 (2) \text{ K}$
$V = 1767.1 (3) \text{ \AA}^3$	Block, colourless
$Z = 4$	$0.35 \times 0.33 \times 0.28 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	2553 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.050$
Monochromator: graphite	$\theta_{\text{max}} = 26.0^\circ$
$T = 293(2) \text{ K}$	$\theta_{\text{min}} = 2.2^\circ$
φ and ω scans	$h = -9 \rightarrow 13$
Absorption correction: none	$k = -16 \rightarrow 20$
9771 measured reflections	$l = -13 \rightarrow 13$
3476 independent reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.041$	H-atom parameters constrained
$wR(F^2) = 0.094$	$w = 1/[\sigma^2(F_o^2) + (0.0421P)^2]$
$S = 0.95$	where $P = (F_o^2 + 2F_c^2)/3$
3476 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
253 parameters	$\Delta\rho_{\text{max}} = 0.20 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$
	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations

between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.45581 (11)	0.01749 (7)	0.14315 (11)	0.0365 (3)
O2	0.07744 (11)	0.13152 (8)	0.15937 (12)	0.0437 (4)
O3	0.74820 (11)	0.10013 (7)	0.38688 (11)	0.0322 (3)
O4	0.34849 (12)	0.24241 (7)	0.14859 (13)	0.0411 (3)
N1	0.28632 (13)	0.07596 (8)	0.18964 (12)	0.0264 (3)
N2	0.53713 (12)	0.16715 (8)	0.29363 (13)	0.0257 (3)
C1	0.34000 (17)	0.04363 (10)	0.10366 (16)	0.0277 (4)
C2	0.14769 (16)	0.10141 (10)	0.11178 (16)	0.0296 (4)
C3	0.66539 (16)	0.14316 (10)	0.29978 (16)	0.0252 (4)
C4	0.46318 (17)	0.21605 (10)	0.17971 (16)	0.0296 (4)
C11	0.36534 (16)	0.09109 (10)	0.33430 (15)	0.0250 (4)
C12	0.48509 (16)	0.13677 (10)	0.38461 (16)	0.0251 (4)
C13	0.55851 (17)	0.15123 (10)	0.52504 (16)	0.0312 (4)
H13A	0.6389	0.1817	0.5587	0.037*
C14	0.51259 (18)	0.12054 (11)	0.61507 (17)	0.0339 (4)
H14A	0.5619	0.1304	0.7092	0.041*
C15	0.39354 (18)	0.07529 (11)	0.56535 (16)	0.0336 (4)
H15A	0.3629	0.0544	0.6261	0.040*
C16	0.31966 (17)	0.06088 (10)	0.42531 (16)	0.0301 (4)
H16A	0.2389	0.0307	0.3921	0.036*
C21	0.22647 (16)	0.05014 (10)	-0.03928 (15)	0.0268 (4)
C22	0.11188 (16)	0.08397 (10)	-0.03443 (16)	0.0283 (4)
C23	-0.00825 (18)	0.09885 (11)	-0.15363 (17)	0.0352 (4)
H23A	-0.0856	0.1211	-0.1505	0.042*
C24	-0.00952 (18)	0.07944 (11)	-0.27824 (17)	0.0374 (4)
H24A	-0.0890	0.0892	-0.3602	0.045*
C25	0.10523 (18)	0.04581 (11)	-0.28296 (17)	0.0388 (5)
H25A	0.1015	0.0332	-0.3680	0.047*
C26	0.22624 (18)	0.03050 (11)	-0.16251 (16)	0.0347 (4)
H26A	0.3037	0.0079	-0.1652	0.042*
C31	0.67265 (16)	0.18160 (10)	0.18068 (15)	0.0270 (4)
C32	0.55298 (16)	0.22536 (10)	0.10967 (16)	0.0295 (4)
C33	0.53101 (18)	0.26772 (11)	-0.00715 (18)	0.0392 (5)
H33A	0.4505	0.2973	-0.0546	0.047*
C34	0.63355 (19)	0.26451 (12)	-0.05105 (18)	0.0429 (5)
H34A	0.6214	0.2921	-0.1298	0.051*
C35	0.75357 (19)	0.22099 (11)	0.02044 (18)	0.0388 (5)
H35A	0.8211	0.2203	-0.0107	0.047*

supplementary materials

C36	0.77526 (17)	0.17830 (11)	0.13755 (17)	0.0327 (4)
H36A	0.8556	0.1487	0.1851	0.039*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0277 (7)	0.0500 (8)	0.0315 (6)	0.0109 (6)	0.0132 (5)	-0.0002 (6)
O2	0.0273 (7)	0.0680 (10)	0.0370 (7)	0.0069 (6)	0.0156 (6)	-0.0122 (6)
O3	0.0240 (6)	0.0402 (8)	0.0282 (6)	0.0050 (6)	0.0082 (5)	0.0051 (5)
O4	0.0267 (7)	0.0422 (8)	0.0558 (8)	0.0104 (6)	0.0199 (6)	0.0156 (6)
N1	0.0210 (7)	0.0358 (8)	0.0233 (7)	0.0002 (6)	0.0107 (6)	-0.0034 (6)
N2	0.0197 (7)	0.0312 (8)	0.0268 (7)	0.0019 (6)	0.0112 (6)	0.0054 (6)
C1	0.0277 (9)	0.0297 (10)	0.0280 (9)	0.0000 (8)	0.0145 (8)	-0.0010 (7)
C2	0.0228 (9)	0.0361 (11)	0.0305 (9)	-0.0017 (8)	0.0124 (8)	-0.0034 (8)
C3	0.0198 (8)	0.0288 (10)	0.0254 (8)	-0.0027 (7)	0.0087 (7)	-0.0018 (7)
C4	0.0247 (9)	0.0293 (10)	0.0339 (9)	0.0008 (8)	0.0123 (8)	0.0039 (8)
C11	0.0224 (9)	0.0304 (10)	0.0223 (8)	0.0031 (7)	0.0101 (7)	-0.0011 (7)
C12	0.0225 (8)	0.0281 (9)	0.0260 (8)	0.0024 (7)	0.0122 (7)	0.0019 (7)
C13	0.0247 (9)	0.0353 (11)	0.0299 (9)	-0.0026 (8)	0.0090 (8)	-0.0035 (8)
C14	0.0342 (10)	0.0432 (12)	0.0220 (8)	0.0019 (9)	0.0107 (8)	-0.0014 (8)
C15	0.0367 (10)	0.0413 (11)	0.0286 (9)	0.0010 (9)	0.0197 (8)	0.0039 (8)
C16	0.0267 (9)	0.0343 (11)	0.0331 (9)	-0.0029 (8)	0.0168 (8)	-0.0006 (8)
C21	0.0251 (9)	0.0294 (10)	0.0252 (8)	0.0009 (7)	0.0108 (7)	-0.0002 (7)
C22	0.0244 (9)	0.0328 (10)	0.0273 (9)	-0.0019 (8)	0.0113 (7)	-0.0035 (7)
C23	0.0247 (9)	0.0423 (12)	0.0329 (9)	0.0027 (8)	0.0080 (8)	-0.0035 (8)
C24	0.0338 (10)	0.0401 (11)	0.0276 (9)	0.0002 (9)	0.0043 (8)	0.0007 (8)
C25	0.0441 (11)	0.0454 (12)	0.0243 (9)	0.0004 (9)	0.0131 (9)	-0.0043 (8)
C26	0.0366 (10)	0.0402 (11)	0.0294 (9)	0.0046 (9)	0.0167 (8)	-0.0024 (8)
C31	0.0240 (9)	0.0300 (10)	0.0271 (8)	-0.0024 (7)	0.0117 (7)	-0.0002 (7)
C32	0.0245 (9)	0.0335 (10)	0.0309 (9)	-0.0004 (8)	0.0127 (8)	0.0041 (8)
C33	0.0325 (10)	0.0434 (12)	0.0395 (10)	0.0044 (9)	0.0143 (9)	0.0137 (9)
C34	0.0461 (12)	0.0509 (13)	0.0350 (10)	-0.0047 (10)	0.0214 (9)	0.0104 (9)
C35	0.0397 (11)	0.0485 (13)	0.0373 (10)	-0.0071 (9)	0.0253 (9)	-0.0013 (9)
C36	0.0278 (9)	0.0378 (11)	0.0340 (9)	-0.0005 (8)	0.0152 (8)	-0.0010 (8)

Geometric parameters (\AA , $^\circ$)

O1—C1	1.2063 (18)	C15—H15A	0.9300
O2—C2	1.2050 (18)	C16—H16A	0.9300
O3—C3	1.2083 (18)	C21—C26	1.377 (2)
O4—C4	1.2101 (18)	C21—C22	1.384 (2)
N1—C1	1.4110 (19)	C22—C23	1.380 (2)
N1—C2	1.412 (2)	C23—C24	1.386 (2)
N1—C11	1.4320 (18)	C23—H23A	0.9300
N2—C4	1.400 (2)	C24—C25	1.383 (2)
N2—C3	1.4131 (19)	C24—H24A	0.9300
N2—C12	1.4328 (19)	C25—C26	1.393 (2)
C1—C21	1.485 (2)	C25—H25A	0.9300
C2—C22	1.484 (2)	C26—H26A	0.9300

C3—C31	1.480 (2)	C31—C32	1.382 (2)
C4—C32	1.484 (2)	C31—C36	1.383 (2)
C11—C16	1.385 (2)	C32—C33	1.381 (2)
C11—C12	1.388 (2)	C33—C34	1.388 (2)
C12—C13	1.386 (2)	C33—H33A	0.9300
C13—C14	1.380 (2)	C34—C35	1.383 (2)
C13—H13A	0.9300	C34—H34A	0.9300
C14—C15	1.378 (2)	C35—C36	1.387 (2)
C14—H14A	0.9300	C35—H35A	0.9300
C15—C16	1.383 (2)	C36—H36A	0.9300
C1—N1—C2	111.16 (13)	C26—C21—C22	121.72 (15)
C1—N1—C11	125.16 (13)	C26—C21—C1	129.55 (15)
C2—N1—C11	123.21 (12)	C22—C21—C1	108.68 (13)
C4—N2—C3	111.54 (12)	C23—C22—C21	121.09 (15)
C4—N2—C12	125.22 (12)	C23—C22—C2	130.42 (15)
C3—N2—C12	122.86 (13)	C21—C22—C2	108.45 (14)
O1—C1—N1	125.04 (14)	C22—C23—C24	117.67 (16)
O1—C1—C21	129.19 (15)	C22—C23—H23A	121.2
N1—C1—C21	105.77 (13)	C24—C23—H23A	121.2
O2—C2—N1	124.91 (15)	C25—C24—C23	121.19 (16)
O2—C2—C22	129.15 (15)	C25—C24—H24A	119.4
N1—C2—C22	105.93 (13)	C23—C24—H24A	119.4
O3—C3—N2	124.90 (14)	C24—C25—C26	121.09 (16)
O3—C3—C31	129.57 (14)	C24—C25—H25A	119.5
N2—C3—C31	105.53 (13)	C26—C25—H25A	119.5
O4—C4—N2	124.59 (15)	C21—C26—C25	117.24 (16)
O4—C4—C32	129.63 (15)	C21—C26—H26A	121.4
N2—C4—C32	105.77 (13)	C25—C26—H26A	121.4
C16—C11—C12	119.44 (14)	C32—C31—C36	121.38 (15)
C16—C11—N1	119.27 (14)	C32—C31—C3	108.62 (13)
C12—C11—N1	121.28 (13)	C36—C31—C3	130.00 (15)
C13—C12—C11	119.99 (14)	C33—C32—C31	121.41 (15)
C13—C12—N2	119.17 (14)	C33—C32—C4	130.07 (15)
C11—C12—N2	120.81 (13)	C31—C32—C4	108.52 (14)
C14—C13—C12	120.17 (15)	C32—C33—C34	117.47 (16)
C14—C13—H13A	119.9	C32—C33—H33A	121.3
C12—C13—H13A	119.9	C34—C33—H33A	121.3
C15—C14—C13	119.96 (15)	C35—C34—C33	121.09 (16)
C15—C14—H14A	120.0	C35—C34—H34A	119.5
C13—C14—H14A	120.0	C33—C34—H34A	119.5
C14—C15—C16	120.11 (15)	C34—C35—C36	121.35 (16)
C14—C15—H15A	119.9	C34—C35—H35A	119.3
C16—C15—H15A	119.9	C36—C35—H35A	119.3
C15—C16—C11	120.33 (15)	C31—C36—C35	117.31 (16)
C15—C16—H16A	119.8	C31—C36—H36A	121.3
C11—C16—H16A	119.8	C35—C36—H36A	121.3

supplementary materials

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C13—H13A \cdots O4 ⁱ	0.93	2.39	3.321 (2)	179
C24—H24A \cdots O3 ⁱⁱ	0.93	2.51	3.405 (2)	161
C26—H26A \cdots O1 ⁱⁱⁱ	0.93	2.53	3.434 (2)	164
C34—H34A \cdots O2 ^{iv}	0.93	2.48	3.410 (2)	177
C36—H36A \cdots O2 ^v	0.93	2.55	3.255 (2)	133

Symmetry codes: (i) $x+1/2, -y+1/2, z+1/2$; (ii) $x-1, y, z-1$; (iii) $-x+1, -y, -z$; (iv) $x+1/2, -y+1/2, z-1/2$; (v) $x+1, y, z$.

Fig. 1

